H. U. Koch and W. Fischer,* Volume 17, Number 24, November 28, 1978, pages 5275–5281.

Page 5275. In the abstract, lines 3–5 should read as follows: identified as 1,2-di-*O*-acyl-3-*O*-[*O*-α-D-glucopyranosyl-(1→2)-α-D-glucopyranosyl]glycerol and 1,2-di-*O*-acyl-3-*O*-[*O*-α-D-glucopyranosyl-(1→2)-(6-*O*-acyl-α-D-glucopyranosyl)]glycerol.

Structure and Thermodynamic Properties of the Complexes between Phospholipase A_2 and Lipid Micelles, by P. Soares de Araujo,* M. Y. Rosseneu, J. M. H. Kremer, E. J. J. van Zoelen, and G. H. de Haas, Volume 18, Number 4, February 20, 1979, pages 580–586.

Page 583. In Table II, for $K'_D \times 10^3$, read $K'_D \times 10^6$.

Measurement of Macromolecular Equilibrium Binding Constants by a Sucrose Gradient Band Sedimentation Method. Application to Protein–Nucleic Acid Interactions, by David E. Draper and Peter H. von Hippel,* Volume 18, Number 5, March 6, 1979, pages 753–760.

Page 755. In Table I, footnote *a*, the NaCl concentration of the buffer used should have been given as 10.0 mM, *not* of 1.0 mM NaCl as listed. This is important because the DNA–RNase binding constant is strongly dependent on salt concentration. In addition, the $Mg(OAc)_2$ concentration listed in Table I, footnote *c*, should have been 5.0 mM.

Investigation of the Pre-Steady-State Kinetics of Fructose Bisphosphatase by Employment of an Indicator Method, by Patricia A. Benkovic, Mohammed Hegazi, Brian A. Cunningham, and Stephen J. Benkovic,* Volume 18, Number 5, March 6, 1979, pages 830–835.

Page 832. In Table I, the buffer capacities of phenol red and FBPase should be 0.066 and 6.87, respectively.

Partial Purification and Characterization of a Human 3-Methyladenine–DNA Glycosylase, by Thomas P. Brent, Volume 18, Number 5, March 6, 1979, pages 911–920.

Page 912. In column 2, lines 26–28 should read as follows: An alternative solvent system consisted of 2-propanol–NH_4OH–H_2O (7:1:2 by volume).

Amino Acid Catalyzed Condensation of Purines and Pyrimidines with 2-Deoxyribose, by Gary L. Nelsestuen, Volume 18, Number 13, June 26, 1979, pages 2843–2846.

An important reference to the scientific literature was inadvertently omitted. Dr. J. A. Carbon [(1964) *J. Am. Chem. Soc. 86*, 720] previously demonstrated the direct condensation of certain purines with deoxyribose. The mechanism is apparently similar, although amine catalysts were not used. Somewhat higher temperatures and longer reaction times will affect the condensation of at least the purines (and, presumably, the pyrimidines) with deoxyribose in the absence of amine catalysts.

Fluorescence Studies of the Pyruvate Dehydrogenase Multienzyme Complex from *Escherichia coli*, by Kimon J. Angelides and Gordon G. Hammes,* Volume 18, Number 7, April 3, 1979, pages 1223–1229.

Because of an error in the calculation of the overlap integral, the distances between pyrene-labeled lipoic acid and FAD are somewhat shorter than in Table II. The error was due to the use of the difference spectrum between the enzyme complex and the enzyme complex free of FAD rather than the difference spectrum between the enzyme complex with oxidized and reduced FAD. This does not alter the conclusions reached. The range in intermolecular distances between *N*-(3-pyrene)maleimide on lipoic acid in different environments and FAD is now 21 to >41 Å rather than 23 to >47 Å. The corrected entries in Table II are given below; FAD is the energy acceptor in all cases.

energy donor	R_0 (Å)	R (Å)
1.5 MalPy	26.6	33
5 MalPy	27.5	34
11 MalPy	26.7	36
13 MalPy	25.5	38
18 MalPy	25.1	>41
35 MalPy	24.8	>41
48 MalPy	23.8	>39
40 MalNEt, 5 MalPy	24.9	21
20 MalNEt, 5 MalPy	24.4	27

Nucleosome Structure: Sites of Interaction of Proteins in the DNA Grooves as Determined by Raman Scattering, by D. C. Goodwin, J. Vergne, J. Brahms,* N. Defer, and J. Kruh, Volume 18, Number 10, May 15, 1979, pages 2057–2064.

Page 2062. Figure 8 is incorrect. The correct figure is